

Combination of ozonation and photocatalysis for pharmaceutical wastewater treatment

by Enjarlis Enjarlis

Submission date: 28-Jul-2022 01:14AM (UTC-0400)

Submission ID: 1876084973

File name: Enjarlis_Treatmen_Pharmaceutical_wastewater_Conference_2017.pdf (728.23K)

Word count: 3750

Character count: 20211

20

See discussions, stats, and author profiles for this publication at: <https://www.researchgate.net/publication/321338152>

Combination of ozonation and photocatalysis for pharmaceutical wastewater treatment

3

Conference Paper in *AIP Conference Proceedings* · November 2017

DOI:10.1063/1.5011876

CITATIONS

2

READS

338

3 authors:



Ratnawati Ratnawati

Institut Teknologi Indonesia

14 PUBLICATIONS 94 CITATIONS

[SEE PROFILE](#)

E. Enjarlis

Institut Teknologi Indonesia

22 PUBLICATIONS 34 CITATIONS

[SEE PROFILE](#)

Slamet Slamet

University of Indonesia

110 PUBLICATIONS 794 CITATIONS

[SEE PROFILE](#)

11

Some of the authors of this publication are also working on these related projects:



C.I. Chemistry and Catalysis Research Grants [View project](#)



Biopolyol synthesis of Tallow by ozonation process [View project](#)

All content following this page was uploaded by Ratnawati Ratnawati on 28 July 2018.

The user has requested enhancement of the downloaded file.

Combination of ozonation and photocatalysis for pharmaceutical wastewater treatment

Ratnawati, Enjarlis, and Slamet

14

Citation: *AIP Conference Proceedings* **1904**, 020019 (2017);

View online: <https://doi.org/10.1063/1.5011876>

View Table of Contents: <http://aip.scitation.org/toc/apc/1904/1>

Published by the American Institute of Physics

Articles you may be interested in

Isolation and characterization of α -cellulose from blank bunches of palm oil and dry jackfruit leaves with alkaline process NaOH continued with bleaching process H_2O_2

AIP Conference Proceedings **1904**, 020001 (2017); 10.1063/1.5011858

18

Application of Taguchi optimisation of electro metal – electro winning (EMEW) for nickel metal from laterite

AIP Conference Proceedings **1904**, 020004 (2017); 10.1063/1.5011861

19

Application of Taguchi optimization on the cassava starch wastewater electrocoagulation using batch recycle method

AIP Conference Proceedings **1904**, 020006 (2017); 10.1063/1.5011863

24

zymatic hydrolysis of lignocellulosic biomass by *Kitasatospora* sp. to produce xylo-oligosaccharides (XOS)

AIP Conference Proceedings **1904**, 020021 (2017); 10.1063/1.5011878

16

Supercritical fluid extraction of ginger (*Zingiber Officinale* Var. *Amarum*) : Global yield and composition study

AIP Conference Proceedings **1904**, 020017 (2017); 10.1063/1.5011874

2

Characterization of biomasses, concentrates, and permeates of dried powder of Kombucha fermentation of spinach (*Amaranthus* sp.) and broccoli (*Brassica oleracea*) with membrane microfiltration and freeze drying techniques for natural sources of folic acid

AIP Conference Proceedings **1904**, 020023 (2017); 10.1063/1.5011880

Combination of Ozonation and Photocatalysis for Pharmaceutical Wastewater Treatment

Ratnawati^{1a)}, Enjarlis^{1b)}, Slamet^{2c)}

17

¹Department of Chemical Engineering, Institut Teknologi Indonesia, Tangerang Selatan, 15320, Indonesia,
²Department of Chemical Engineering, Faculty of Engineering, Universitas Indonesia, Depok 16424, Indonesia.

45

^{a)}Corresponding author email: rnwt63@yahoo.co.id

^{b)}enjarlis@yahoo.com

^{c)}slamet@che.ui.ac.id

Abstract. The chemical oxygen demand (COD) and phenol removal from pharmaceutical wastewater were investigated using configuration of two circulation batch reactors in a series with ozonation and photocatalytic processes. The ozonation is conducted with O₃/granulated activated carbon (O₃/GAC), whereas photocatalysis with TiO₂ that immobilized on pumice stone (PS-TiO₂). The effect of circulation flow rate (10; 12; 15 L/min) and the amount PS-TiO₂ (200 g, 250 g, 300 g) were examined. Wastewater of 20 L was circulated pass through the pipe that injected with O₃ by the ozone generator, and subsequently flow through two GAC columns, and finally, go through photoreactor that contains photocatalyst PS-TiO₂ which equipped with mercury lamp as a photon source. At a time interval, COD and phenol concentration were measured to assess the performance of the process. FESEM imaging confirmed that O₂ was successfully impregnated on PS, as corroborated by EDX spectra. Meanwhile, degradation process indicated that the combined ozonation and photocatalytic processes (O₃/GAC-TiO₂) is more efficient compared to the ozonation and photocatalysis alone. For combination process with the circulation flow rate of 10 L/min and 300 g of PS-TiO₂, the influent COD of around 1000 ppm are effectively degraded to a final effluent COD of 290 ppm (71% removal) and initial phenol concentration of 4.75 ppm down to 0 ppm for 4 h which this condition fulfill the discharge standards quality. Therefore, this portable prototype reactor is effective that can be used in the pharmaceutical wastewater treatment. For the future, this process condition will be developed for orientation on the industrial applications (portable equipment) since pharmaceutical industries produce wastewater relatively in the small amount.

INTRODUCTION

Nowadays, the environmental problems are faced by countries in the world. The pharmaceutical industry is one of the most significant sectors in the world's economy especially in the countries that have big population. The increasing demand of medicine gives the positive impact on economic. On the other hand, it also creates the negative impact on environmental problems. High concentration pharmaceutical industry wastewater needs the high degree of treatment before it discharged into the environment [1]. Since it categorized as harmful wastewater, conventional treatment such as physical, chemical, and biological methods are not adequate to treat this waste [2]. Recently, Advanced Oxidation Processes (AOPs) such as ozonation [3,4] and photocatalysis [5] are wastewater treatments that provide the high removal of pollutants. In ozone process, the availability of O₃ and OH radicals that have the high oxidative power play an important role in the degradation of pollutants. For instance, high efficiency of ozone process for drinking water system has been studied [6], however, ozone process is relatively expensive due to the consumption of electricity. Meanwhile, photocatalysis with semiconductor TiO₂ is a green technology for degrading wide variety organic pollutants with solar energy [4,5], and other environmental application such as air quality control, CO₂ removal and disinfectant of bacteria [7,8]. Moreover, semiconductor TiO₂ is an effective photocatalyst due to the thermal stability, low toxicity, noncorrosive and inexpensive [9]. Photocatalysis with TiO₂ impregnated on pumice provides some advantages since pumice can act as adsorbent and the capability of pumice to float improves photon absorption. In addition, TiO₂ impregnated on the pumice also reduce the difficulty in the filtration of the photocatalyst since filtration

5

Proceedings of the 3rd International Symposium on Applied Chemistry 2017
 AIP Conf. Proc. 1904, 020019-1–020019-8; <https://doi.org/10.1063/1.5011876>
 Published by AIP Publishing. 978-0-7354-1594-2/\$30.00

can be eliminated by immobilizing TiO_2 on the solid support such as pumice [5,10]. However, photocatalysis still has limitation since its low oxidation rate, only suitable for the dilute wastewater and in some cases the complete mineralization is slowly [4]. In general, a combination of several treatment methods provides the high efficiency removal of pollutants compared with individual method [11]. Therefore, the combination of ozone process and photocatalysis is reasonable for treatment pharmaceutical wastewater that indicated the high concentration of pollutants. This process is required in order to fulfill the discharge standards quality in short time treatment process. Actually, some researchers have investigated this combination, however, they used synthetic and single waste such as formic acids, aniline, pyridine, humic acid etc [4]. To the best of our knowledge, the application of this combination has not been investigated yet for the real wastewater especially pharmaceutical wastewater that consists of hardly degradable substances indicated by the high concentration of COD, BOD, TDS, and phenol.

EXPERIMENTAL

Material and Methods

The pharmaceutical wastewater used in this study was obtained from PT X Indonesia located in Tangerang. The effluent has pH around 5 – 6 with the COD value around 1000 ppm and the initial phenol concentration was 4.75 ppm. The ozone was generated from the air by the ozone generator (Hanaco THS-278 type) with the dosage of 0.1 g/h. This dosage was obtained by Iodometric titration method. Granular activated carbon was purchased and it was applied without any further treatment. Commercial powder of TiO_2 P25 (79% anatase, 21% rutile) was supplied from Evonik Industries. Pumice is supplied from Eka City, Nusa Tenggara Barat, Indonesia. Tetra Ethyl Ortho Silicate (TEOS) or $\text{Si}(\text{OC}_2\text{H}_5)_4$ 98% and all chemicals used in this study were purchased from Sigma-Aldrich.

TiO_2 immobilized on pumice was performed using impregnation method. Pumice (PS) with the size of 4-6 mm was cleaned ultrasonically. Dip coating of PS in the sol of TiO_2 (5 % w/w of TiO_2 on PS) was performed with TEOS as an adhesive and Si source. Finally, the composite PS- TiO_2 was evaporated in the furnace at 350 °C for 1 h. The PS- TiO_2 was characterization by FESEM/EDX (FEI inspect F50) and then it underwent photocatalyst test for degrading pharmaceutical wastewater.

Batch-recirculation reactor system and wastewater treatment procedure.

The configuration of two batch circulation reactors with ozonation and photocatalytic processes can be seen in Fig. 1. The waste from the wastewater tank that made of glass with the volume of 30 L (length 30 cm, width 30 cm, and height 34 cm) was firstly pumped through the pipe (ID = 2.1 cm) and it was injected with O_3/O_2 from the ozone generator with the O_3 producing rate 0.1 g/h. Subsequently, the waste went through the two carbon columns and finally, it experienced photodegradation in the 30 L of the photoreactor that equipped with the mercury lamp (HPL-N 250 W/542 E40 HGISL) as a photon source (17% of UV and 83% of visible light). For the ozonation, the schematic process was similar to the Figure 1, however, the flow after through GAC was directly recirculation to wastewater tank. For the photocatalysis, the wastewater was allowed to flow directly to photoreactor and back to the wastewater tank. The circulation flow rate of the wastewater was setup and varied (10; 12; 15) L/min, whereas the amount of the PS- TiO_2 also varied (200; 250; 300) g. At a time interval, the COD and phenol concentration were measured according to SNI and ASTM standards.

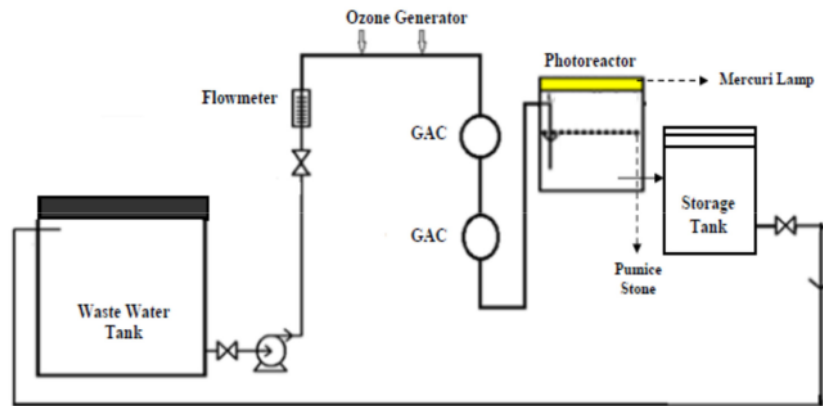


FIGURE 1. The experimental setup for ozone process and photocatalysis

RESULT AND DISCUSSION

Photocatalyst characterization

The FESEM image of PS-TiO₂ indicated that PS has pores and the surface has covered by TiO₂ as shown in Fig. 2.

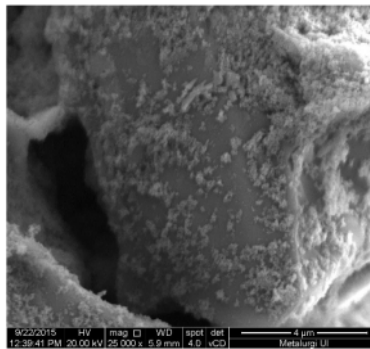


FIGURE 2. FESEM image of PS-TiO₂ (magnification of 25000 X)

The EDX results of PS-TiO₂ (Table 1 and Fig. 3) indicated that besides O and Ti, the components like Na, Al, Si, K, and Ca that contained in PS were detected and it gave evident the successful impregnation of TiO₂ on PS. Similar result also reported by previous researchers [10,12,13]. Table 1 presents the elemental composition of PS-TiO₂.

Table 1. Elemental composition of PS-TiO₂

Elements	%Weight
O	63.1
Na	2.7
Al	3.69
Si	11.91
K	0.63
Ca	0.33
Ti	17.2

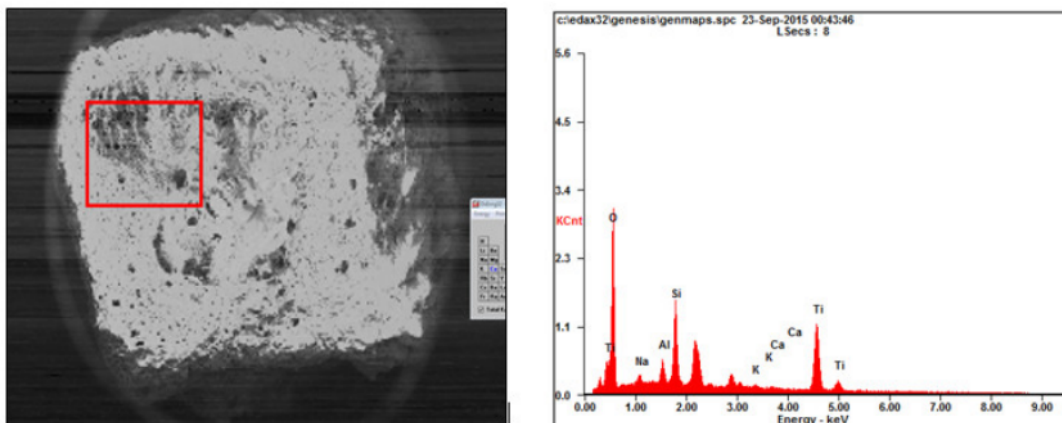


FIGURE 3: EDX analysis of PS-TiO₂

Pollutants degradation by ozone process

Figure 4 shows the reduction of COD from wastewater by O₃/GAC as a function of time with 300 g of PS-TiO₂ at the different circulation flow rate. The reduction of COD is caused by the oxidation of the pollutant with: a). O₃ (ozonolysis), b). with hydroxyl radical (\bullet OH) results from the reaction of O₃ with water, and c). the reaction of O₃ in the surface of GAC. The function of GAC is not only as the adsorbent that removes organic carbon, but also accelerate the O₃ transformation into \bullet OH [5]. As a result, ozonation process is very effective in the degradation of pollutant in wastewater. The high performance of ozonation is caused by the high oxidative power of hydroxyl radical and ozone with the redox potential of 2.8 V and 2.07 V respectively [1,4]. There are two reactions through which O₃ decomposes the pollutant. Firstly, direct attack of pollutants by molecular ozone (ozonolysis) that occurs at pH below 7 (acidic or neutral condition). This condition is the selective reaction that results in the end product that can't be oxidized further by O₃. Secondly, nonselective decomposition of pollutants by OH radicals at high pH value [4]. It is clear that the circulation flow rate of 10 L/min of the wastewater provides the highest COD removal since at this condition the contact time and the mixing of O₃ and \bullet OH with the pollutant in wastewater are sufficient and better compared to others. As a result, pollutant could be degraded effectively. This result is in accordance with the previously reported by Linda Zou et al. (2008) when they studied color removal from reused water with ozone and photocatalytic process [14]. To fulfill the COD standard quality requirement (below 300 ppm), this process need around 12.6 h. For phenol degradation, it diminished from 4.75 ppm to become 0.1 ppm in 6 h ozonation process. On the other hand, at higher circulation flow rate, the contact time of ozone and hydroxyl radical to the pollutant is decreased, which resulted in the slower COD removal. The reduction of COD and phenol concentration due to the oxidation of pollutant in the pharmaceutical wastewater to become degradation product by O₃, \bullet OH, and adsorbed by GAC.

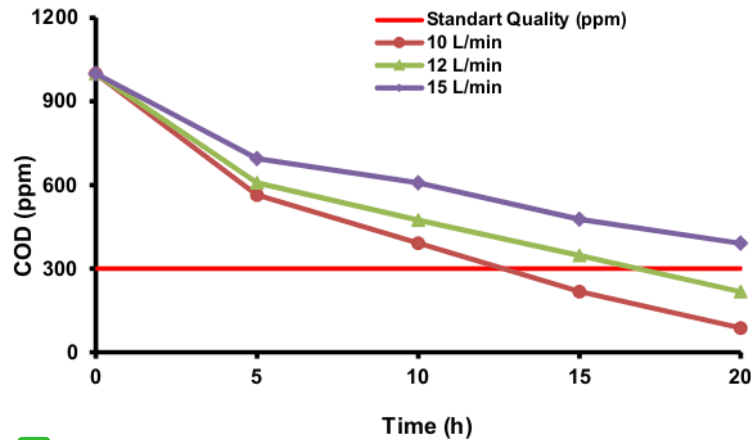
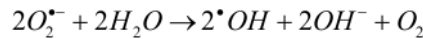
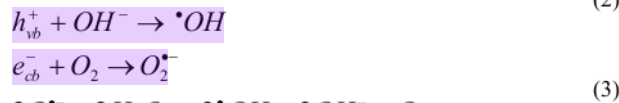
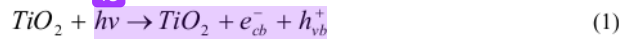


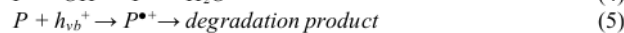
FIGURE 4. The COD reduction as a function of time with ozonation process for various circulation flow rate

Pollutants degradation by photocatalysis

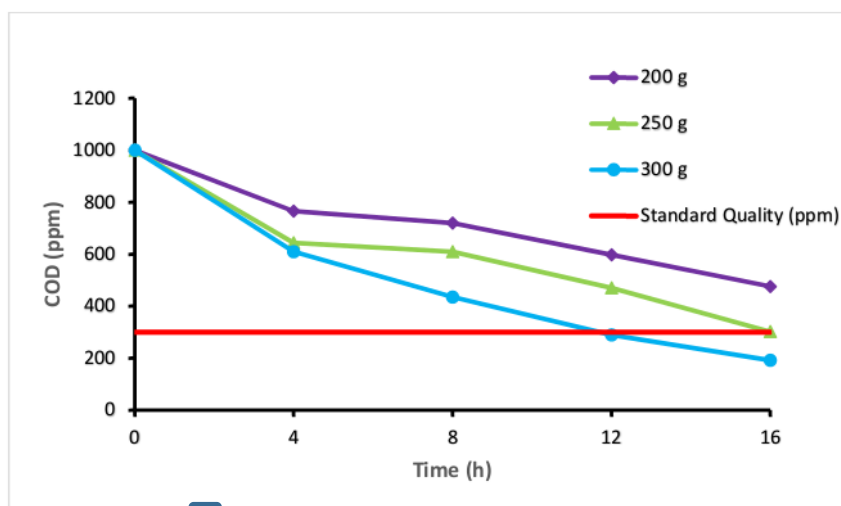
Figure 5 depicts COD removal of pharmaceutical wastewater as a function of time irradiation at 10 L/min circulation flow rate with different amount of photocatalyst. It is clear that the COD rapidly reduced by the photocatalytic process with 300 g of PS-TiO₂ compared to others. In this process, the hydroxyl radical is considered as the main substance that responsible for degradation of organic matters [11, 14]. In photocatalysis, when TiO₂ absorbs photon greater than its bandgap energy, there is excitation electrons from valence band (vb) to conduction band (cb) and therefore, resulting electrons in the conduction band and holes in the valence band according to Eq. 1. The holes react with water to produce •OH (Eq. 2), and this •OH and the holes itself oxidizes the pollutants to become CO₂ and H₂O [4,15]. Hole is the highly oxidizing agent (+2.53 eV versus standard hydrogen electrode (SHE)) that effectively degrade the pollutant [16]. Meanwhile, the electrons can act as the reducing agent to remove heavy metals such as mercury, arsenic, and chromium [17] or reacts with the adsorbed oxygen on the surface of PS-TiO₂ to form hydroxyl radical according to Eq. 3 through various pathways, [4,11]. The reactions in the photocatalysis are as follow.



The pollutants (P) that adsorbed in the surface of PS-TiO₂ plays an important role in the photocatalytic oxidation via successive attacks by •OH or hole as follows [14].



From Fig.5 it can be stated that the amount of PS-TiO₂ significantly affects the COD removal. The more the amount of PS-TiO₂, the higher the COD removal and 300 g of the photocatalyst in this process result in the highest COD removal. This condition is considered as the optimum process. It meant that more •OH and hole are responsible in degrading pollutant. This photocatalysis needs 12 h to fulfill below the COD standard quality requirement (300 ppm) that allowed to discharge in the environment. In this condition, the percentage COD removal was 71 %. To reduce the time of the process, the combination of ozonation and photocatalysis was performed.



44
FIGURE 5. The COD removal as a function of time with photocatalytic process for various PS loading

Pollutants degradation by combination of ozone process and photocatalysis

The combination of ozonation followed by photocatalysis processes with batch circulation reactors was performed with the equipment that can be seen in Fig. 2 at the flow rate of 10 L/min and 300 g of the photocatalyst. The PS-TiO₂ covered the surface of the liquid waste in the photoreactor. From Fig. 6, it is clear that the COD removal was rapidly degraded by the combination of the two methods compared to the degradation by the photocatalysis or ozonation alone. The result indicated that after 4 h circulation, the initial COD and phenol concentration of around 1000 and 4.75 ppm declined to 290 and 0 ppm, respectively. On the other hand, the photocatalysis or ozonation alone only reduces the COD to 609 and 651 ppm, respectively. This result is in a good agreement with the study that previously reported [11,14]. In ozone process, O₃ and hydroxyl radicals are responsible for pollutant decomposition. However, electrons generation by photocatalytic also could react with ozone to product •OH via several pathways [14] as follows.



The availability of O₂ introduced via ozone generator also can react with the electron to become •OH as final products (Eq. 3). As a consequence, a larger number of •OH (powerful oxidant species) can be produced by the combination of those two processes. In another word, the rapid degradation using this combination process due to not only the role of O₃, •OH, GAC in ozonation and the influence of •OH and hole that resulted from photocatalysis process, but also the increasing the number of radical hydroxyls. As a result, the combination process results in more powerful oxidant that responsible for degradation pollutant. The use of combined ozonation and photocatalysis is an attractive route because of the enhancement of the hydroxyl radical generation, a powerful oxidant agent that can oxidize completely the organic matter present in the aqueous system. Ozone process is reported to be efficient for highly contaminated wastewater such as pharmaceutical waste, meanwhile, photocatalysis can completely decompose the intermediate product that resulted from ozone process. Moreover, photocatalysis also found to be efficient to degrade the dilute solution.

Compared to the ozone process, the photocatalytic process indicated slightly higher COD removal. It has been reported that, most of the organic matters could not completely decompose by ozone alone [14]. Moreover, the pH of this treatment was around 5 – 6, and therefore, pollutant was decomposed mainly by direct attack of O₃ and it is a selective reaction that resulted in intermediate product. Since this intermediate product cannot be oxidized further, the COD value was slightly higher [4].

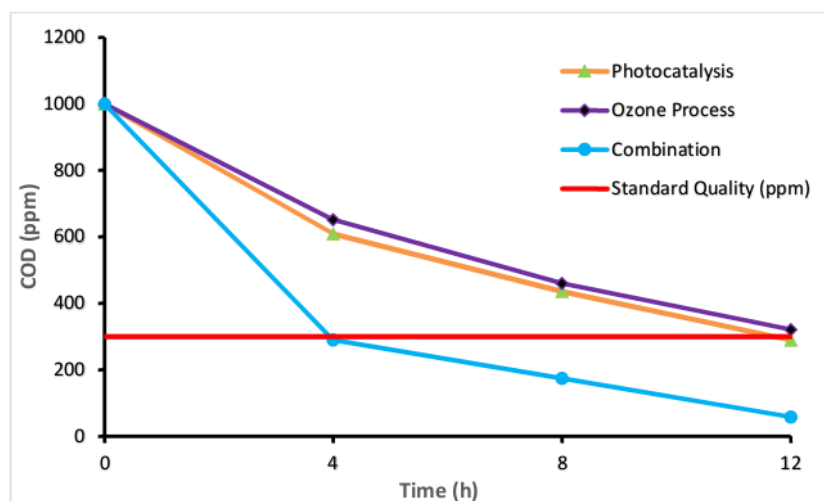


FIGURE 6. The COD removal as a function of time with variation process

CONCLUSION

The combination of ozone process and photocatalysis in a batch reactor with circulation shows as an effective method for pharmaceutical wastewater treatment with relatively high contaminated wastewater indicated by the COD value of 1000 ppm. This process can treat 20 L of pharmaceutical wastewater with the initial COD of 1000 ppm and phenol concentration of 4.75 ppm decrease to 290 ppm and 0 ppm, respectively for 4 h with the circulation flow rate of 10 L/min and 300 g PS-TiO₂. This condition meets the standard quality of effluent and it is considered as an optimum process. Ozone and photocatalytic processes could be an effective and alternative treatment for pharmaceutical wastewater with high concentration of pollutants.

4

ACKNOWLEDGMENTS

The authors would like to thank LP3M Institut Teknologi Indonesia and Directorate General of Higher Education (DGHE), Indonesian Ministry of National Education for the financial support of this research (Hibah Bersaing grant no. 71/SP/LP3M/ITI/III/2015). Thank also to Dyno Kafaerli Cearsario, Rouhan Hadiwarna and M. Andi Ardiansyah for the data collection.

REFERENCES

1. K. Shalini, Z. Anwer, P. K. Sharma, A Review on Pharma Pollution, *Int. J. of PharmTech Research* 2, 2265-2270 (2019).
2. N. Gedam, N. N. Rao, Carbon attrition during continuous electrolysis in carbon bed based three-phase three-dimensional electrode reactor: Treatment of recalcitrant chemical industry wastewater, *J. Environ. Chem. Eng.* 2, issue 3, 1527-1532 (2014).
3. L. Bilinska, M. Gmurek, S. Ledakowicz, Comparison between industrial and simulated textile wastewater treatment by AOPs-Biodegradability, toxicity, and cost assessment, *Chem. Eng. J.* 306, 550-559 (2016).
4. T. E. Agustina, H. M. Ang, V. K. Vareek, A review of synergistic effect of photocatalysis and ozonation on wastewater treatment, *J. Photochem. Photobiol. C: Photochemistry Reviews* 6, 264-273 (2005).
5. S. Souzanchi, F. Vahabzadeh, S. Fazel, S. N. Hosseini, Performance of an Annular Sieve –Plate Column photoreactor using immobilized TiO₂ on stainless steel support for phenol degradation, *Chem. Eng. J.* 223, 268-276 (2013).
6. M. S. Polo, E. Salhi, J. R. Utrilla, U. von Gunten, Combination of Ozone with Activated Carbon as an Alternative to Conventional Advanced Oxidation processes. *Ozone: Science and Engineering* 28, 237-245 (2006).

7. Y. Wang , Y. He, Q. Lai, M. Fan, Review of the progress in preparing nano TiO₂: An important environmental engineering material,*J. Environ. Sci.* 26, issue 11, 2139-2177 (2014).
8. W. Wang, G. Huang, J.C. Yu, P. K. Wong, Advances in photocatalytic disinfection of bacteria: Development of photocatalyst and mechanism, *J. Environ. Sci.* 34, 232-247 (2015).
9. Ratnawati, J. Gunlazuardi, E. L. Dewi & Slamet, Effect of NaBF₄ addition on the anodic synthesis of TiO₂ nanotube arrays photocatalyst for production of hydrogen generation from glycerol-water solution,*Int. J. Hydrog. Energy*39, 16927-16935 (2014).
10. M. Subrahmanyam, P. Boule, V. D. Kumari, D. N. Kumar, M. Sancelme, A. Rachel, Pumicestone supported titanium dioxide for removal of pathogen in drinking water and recalcitrant in wastewater, *Solar Energy* 82, 1099-1106 (2008).
11. S. Wang, F. Shiraishi, K. Nakano, A synergistic effect of photocatalysis and ozonation on decomposition of formic acid in an aqueous solution, *Chem. Eng. J.* 87,261-271 (2002).
12. K. V. S. Rao, A. Rachel, M. Subrahmanyam, P. Boule, Immobilization of TiO₂ on pumice stone for the photocatalytic degradation of dyes and dye industry pollutants, *Appl. Catal. B: Environ.*46, 77-85 (2003).
13. K. V. S. Rao, M. Subrahmanyam, P. Boule, Immobilized TiO₂ photocatalyst during long-term use: decrease of its activity, *Appl. Catal. B: Environ.* 49, 239-249 (2004).
14. L. Zhou, B. Zou, The synergistic effect of ozonation and photocatalysis on color removal from reused water, *J. Photochem. Photobiol. A: Chemistry*196, 24-32 (2008).
15. Y. Zhu, Y. Wang, Z. Chen, L. Qin, L. Yang, Z. Zhu, et. al.. Visible light induced photocatalysis on CdS quantum dots decorated TiO₂ nanotube arrays,*Appl. Catal. A: General* 498, 159-166 (2015).
16. A. B. Makama, A. Salmiaton, E. B. Saion, T. S. Y. Choong, N. Abdullah, Synthesis of CdS Sensitized TiO₂ Photocatalysts: Methylene Blue Adsorption and Enhance Photocatalytic Activities. *International Journal of Photoenergy*, Volume Article ID 2947510, 1-14 (2016).
17. Y. Kim, H. Joo, N. Her, Y. Yon, C.H. Lee, J. Yoo, Self-rotating photocatalytic system for aqueous Cr(VI) reduction on TiO₂ nanotube/Ti mesh substrate, *Chem. Eng. J.*229, 66-71 (2013).

Combination of ozonation and photocatalysis for pharmaceutical wastewater treatment

ORIGINALITY REPORT

23%
SIMILARITY INDEX

16%
INTERNET SOURCES

19%
PUBLICATIONS

8%
STUDENT PAPERS

PRIMARY SOURCES

- 1** Linda Zou, Bo Zhu. "The synergistic effect of ozonation and photocatalysis on color removal from reused water", Journal of Photochemistry and Photobiology A: Chemistry, 2008
Publication 1%
 - 2** journal.ugm.ac.id
Internet Source 1%
 - 3** Submitted to Universitas Indonesia
Student Paper 1%
 - 4** www.matec-conferences.org
Internet Source 1%
 - 5** Muhammad Arifuddin Fitriady, Anny Sulaswatty, Egi Agustian, Salahuddin, Deska Prayoga Fauzi Aditama. "Supercritical fluid extraction of ginger (Zingiber Officinale Var. Amarum) : Global yield and composition study", AIP Publishing, 2017
Publication 1%
-

6	Rizka Karima, Evana Yuanita, Bunda Amalia, Agustina Arianita, Tiara Mailisa, Bumiarto Nugroho. "Sintesis Micro-Fibrillated Cellulose dari Serat Tandan Kosong Sawit dengan Hidrolisis Asam Oksalat", JURNAL SELULOSA, 2020	1 %
Publication		
7	Submitted to Case Western Reserve University	1 %
Student Paper		
8	profdoc.um.ac.ir	1 %
Internet Source		
9	Shinpon Wang, Fumihide Shiraishi, Katsuyuki Nakano. "A synergistic effect of photocatalysis and ozonation on decomposition of formic acid in an aqueous solution", Chemical Engineering Journal, 2002	1 %
Publication		
10	Ratnawati, Jarnuzi Gunlazuardi, Slamet. "Photocatalytic Decomposition of Glycerol Solution on TiO ₂ Nanotube Arrays (TNTA) Doped with C and N to Produce Hydrogen", Materials Science Forum, 2017	1 %
Publication		
11	www.coursehero.com	1 %
Internet Source		

12

Internet Source

1 %

13

Agustina, T.E.. "A review of synergistic effect of photocatalysis and ozonation on wastewater treatment", Journal of Photochemistry & Photobiology, C: Photochemistry Reviews, 200512

Publication

1 %

14

Submitted to Universitas Muhammadiyah Surakarta

Student Paper

1 %

15

Submitted to University Tun Hussein Onn Malaysia

Student Paper

1 %

16

www.SciRP.org

Internet Source

1 %

17

Slamet, Ratnawati, Jarnuzi Gunlazuardi, Eniya Listiani Dewi. "Enhanced photocatalytic activity of Pt deposited on titania nanotube arrays for the hydrogen production with glycerol as a sacrificial agent", International Journal of Hydrogen Energy, 2017

Publication

1 %

18

Submitted to Universidad de Santiago de Chile

Student Paper

1 %

eprints.ucm.es

19	Internet Source	1 %
20	nespurban.edu.au Internet Source	1 %
21	www.tandfonline.com Internet Source	1 %
22	mountainscholar.org Internet Source	<1 %
23	www.nature.com Internet Source	<1 %
24	Nanik Rahmani, Alifah Mafatikhul Jannah, Puspita Lisdiyanti, Bambang Prasetya, Yopi. "Enzymatic hydrolysis of lignocellulosic biomass by Kitasatospora sp. to produce xylo-oligosaccharides (XOS)", AIP Publishing, 2017 Publication	<1 %
25	cyberleninka.org Internet Source	<1 %
26	Jorge Villaseñor, Patricio Reyes, Gina Pecchi. "Catalytic and photocatalytic ozonation of phenol on MnO ₂ supported catalysts", Catalysis Today, 2002 Publication	<1 %
27	vuir.vu.edu.au Internet Source	<1 %

28

Sara Moeini Najafabadi, Fariborz Rashidi, Mohammad rezaei. "Performance of a multistage rotating mesh support photoreactor immobilized with TiO₂ on photocatalytic degradation of PNP: Reactor construction and optimization", Chemical Engineering and Processing - Process Intensification, 2019

Publication

<1 %

29

Seong-Nam Nam, Hyekyung Cho, Jonghun Han, Namguk Her, Jaekyung Yoon. "Photocatalytic degradation of acesulfame K: Optimization using the Box-Behnken design (BBD)", Process Safety and Environmental Protection, 2018

Publication

<1 %

30

Teh, Swe Jyan, Soo Ling Yeoh, Kian Mun Lee, Chin Wei Lai, Sharifah Bee Abdul Hamid, and Kwai Lin Thong. "Effect of reduced graphene oxide-hybridized ZnO thin films on the photoinactivation of Staphylococcus aureus and Salmonella enterica serovar Typhi", Journal of Photochemistry and Photobiology B Biology, 2016.

Publication

<1 %

31

www.hindawi.com

Internet Source

<1 %

32

Xiaobo Li, Jian Liu, Anthony F. Masters, Vishnu K. Pareek, Thomas Maschmeyer. "Hollow micro/nanomaterials as nanoreactors for photocatalysis", APL Materials, 2013

Publication

<1 %

33

ethesis.nitrkl.ac.in

Internet Source

<1 %

34

repositorio-aberto.up.pt

Internet Source

<1 %

35

scholar.ui.ac.id

Internet Source

<1 %

36

Bo Zhu, Linda Zou. "Trapping and decomposing of color compounds from recycled water by TiO₂ coated activated carbon", Journal of Environmental Management, 2009

Publication

<1 %

37

Nitin Gedam, Nageswara Rao Neti. "Carbon attrition during continuous electrolysis in carbon bed based three-phase three-dimensional electrode reactor: Treatment of recalcitrant chemical industry wastewater", Journal of Environmental Chemical Engineering, 2014

Publication

<1 %

38

Venkata Subba Rao, K.. "Immobilization of TiO₂ on pumice stone for the photocatalytic

<1 %

degradation of dyes and dye industry pollutants", Applied Catalysis B, Environmental, 20031031

Publication

39

Wang, Jenny. "Ozone/H₂O₂ AOP pilot testing at a waterworks with dynamic raw water characteristics in Singapore", Water Practice & Technology, 2012.

Publication

<1 %

40

espace.library.uq.edu.au

Internet Source

<1 %

41

notablesdelaciencia.conicet.gov.ar

Internet Source

<1 %

42

patents.justia.com

Internet Source

<1 %

43

tudr.thapar.edu:8080

Internet Source

<1 %

44

Gomez-Pacheco, C.V.. "Tetracycline removal from waters by integrated technologies based on ozonation and biodegradation", Chemical Engineering Journal, 20111215

Publication

<1 %

45

Ratnawati, Jarnuzi Gunlazuardi, Slamet. "Synthesis of TiO Nanotube Arrays by Sonication Aided Anodization and Its Application for Hydrogen Generation from

<1 %

Aqueous Glycerol Solution ", MATEC Web of Conferences, 2015

Publication

46

"Waste Bioremediation", Springer Science and Business Media LLC, 2018

Publication

<1 %

47

Anna Kubacka, Marcos Fernández-García, Gerardo Colón. "Advanced Nanoarchitectures for Solar Photocatalytic Applications", Chemical Reviews, 2011

Publication

<1 %

48

Ratnawati, Jarnuzi Gunlazuardi, Eniya Listiani Dewi, Slamet. "Effect of NaBF₄ addition on the anodic synthesis of TiO₂ nanotube arrays photocatalyst for production of hydrogen from glycerol–water solution", International Journal of Hydrogen Energy, 2014

Publication

<1 %

Exclude quotes Off

Exclude matches Off

Exclude bibliography On